1. COURSE DECRYPTION – GENERAL INFORMATION					
1.1. Course teacher	Professor Tihana Žanić Grubišić, PhD Professor Karmela Barišić, PhD Professor Lada Rumora, PhD	1.6. Year of study	2 nd		
1.2. Name of the course	Biochemistry	1.7. Credit value (ECTS)	8.5		
1.3. Associate teachers	Marija Grdić Rajković, PhD Anita Somborac Bačura, PhD; Andrea Hulina, mag. med. biochem.	1.8. Type of instruction (number of hours L+E+S+e-learning)	60+30+10		
 Study programme (undergraduate, graduate, integrated) 	Pharmacy integrated study programme	1.9. Expected enrolment in the course	130		
1.5. Status of the course	Compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2 nd		
2. COURSE DESCRIPTION					
2.1. Course objectives	Student will acquire fundamental knowledge necessary to understand molecular logics of biochemical processes in living organisms; recognise dynamics in the synthesis and degradation of the natural biomacromolecules: proteins, polysaccharides, lipids and nucleic acids, and will be able to analyse and identify important factors that are influencing dynamics, control and regulation of cellular metabolism. Biochemical knowledge and skills acquired are compulsory basis for the further studies, especially in clinical biochemistry, haematology, pharmacology, biochemistry of drug metabolism, nutrition, molecular biology and genetic engineering, molecular diagnostics, identify molecular basis of diseases and therapy and other lessons dealing with metabolism of endogenic macromolecules, drugs and other xenobiotic in health and disease.				
2.2. Enrolment requirements and required entry competences for the course	Enrolment requirements: completed study course Biological Chemistry				
2.3. Learning outcomes at the level of the study programme to which the course contributes	 Students will be able to apply fundamental biochemical knowledge to explain, analyse and evaluate procedures related to the research, development and production of drugs. Development and implementation of the optimal solutions for practical and everyday problems in monitoring progress of the drug therapy (research and application of new laboratory diagnostic procedures for therapeutic 				

	drug monitoring).			
	Critical evaluation and application of the scientific data and expert knowledge for the problem solving in			
	biochemical systems.			
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	After passing the exam students will be able to:			
	1. Apply expert knowledge of biochemistry, chemistry and biology in biochemical problem solving;			
	2. Describe and apply basic biochemical principles for relating structure and function of the protein macromolecules;			
	3. Determine key enzymes regulating reaction rate in the metabolic pathways and asses what diagnostically			
	measurable biochemical changes might indicate disorders in the particular enzyme systems. Estimate what			
	genetic factors might be relevant to diseases and relate enzyme kinetics to regulatory enzyme characteristics;			
	4. Explain biochemical mechanism of the DNA replication, generation and repair of the DNA mutations, recognise			
	role of all elements in the process of transcription and protein synthesis, in prokaryotes and eukaryotes;			
	5. Review basic principles of acquiring and processing of data in pharmacogenetics, transcriptomics and proteomics			
	6. Designing and performing biochemical experiments based on grasped experimental and technical skills;			
	7. Analyse scientific data bases for the interpretation of the personal results and presentation to the professional			
	audience.			
	LECTURES AND SEMINARS:			
	Dynamic aspects of structure and function of specific proteins: haemoglobin, myoglobin, collagen, elastin, proteins of the extracellular matrix			
2.5. Course content broken down in detail by weekly class schedule (syllabus)	Structure and function of cell membranes in various tissues, transport of ions, amino acids, sugars			
	 Methods for exploring proteins Generation, transforming and storing of metabolic energy: basic concepts of metabolism 			
	Glycolysis			
	Oxidative decarboxylation of pyruvate, citric acid cycle Callylar, biconstration, and role of ATP, generation, and expenditure, reconstratory, about and exidative.			
	Cellular bioenergetics and role of ATP generation and expenditure, respiratory chain and oxidative phosphorylation			
	Gluconeogenesis and pentose phosphate pathway			
	Glycogen metabolism, glycogenesis and glycogenolysis, reactions and hormone regulation			
	 Biochemistry of hormones: insulin, epinephrine and cortisol Fatty acid metabolism, degradation and synthesis of triglycerides, biosynthesis and β-oxidation of fatty acid, 			
	biosynthesis of ketone bodies			
	Protein turnover and amino acid catabolism, urea cycle			

	 Biosynthesis of macromolecular precursors, amino acids, ribonucleotides and deoxyribonucleotides Information in biological systems: DNA - structure and genetic role, genome organisation, chromosomes and genes Methods for exploring genome Histones and DNA packing, conformation of DNA molecule, DNA replication, fidelity of replication DNA mutations and repair RNA in translation of genetic message Synthesis and modification of functional RNA molecules: mRNA and transcription, t-RNA, activation and role in protein synthesis, structure of ribosomes and rRNA Genetic code and relation of genes and proteins, protein synthesis and protein sorting Control of gene expression in prokaryotes: Lac-operon and Trp-operon Chromosomes in eukaryotes and control of gene expression in eukaryotes, introns and exons Integration of biochemical processes in the cell - basic concepts and design, strategy, control and regulation of metabolism EXERCISES: Determination of initial velocity v_o in acetylcholine reaction Alkaline phosphatase Homogenisation, differential centrifugation, determination of DNA and lactate in cellular fractions Isolation of plasmid DNA from the transformed bacteria Rate of glycolysis in various tissues Electrophoresis of haemoglobin 					
2.6. Type of instruction	lectures seminars and workshops exercises online in entirety mixed e-learning field work		independent study multimedia and the internet laboratory work with the mentor (other)		2.7. Comments:	
2.8. Student responsibilities	Regular attendance at the lectures, obligatory attendance at the seminars and laboratory practice, active participation in seminars with individual presentations.					
2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to the credit value of the course)	Class attendance Experimental work	0.5 1.5	Research Report		Practical training	
	Essay		Seminar essay		Semestral written tests	1.5
	Tests	0.5	Oral exam	2.5	(Other—describe)	
	Written exam	2.0	Project		(Other—describe)	
2.1. Grading and evaluation of student work over the course of	Seminars are organised as achievements. During the le					

instruction and at a final exam	are organised for laboratory practicals, before and after completing exercises. At the end of the complete lecture		
	program written and oral exam are organised for the whole program and final grade is decided.		
2.2. Required literature (available at the library and via other media)	Title		
	JM Berg, JL Tymoczko, L. Stryer: Biochemistry, 7 th edition, Školska knjiga, Zagreb, 2013.		
	TM Devlin: Texbook of Biochemistry with Clinical Correlation, J. Wiley & sons, New York, 2011.		
2.12. Optional literature	C. Smith, AD Marks: MMarks' basic Medical Biochemistry, A Clinical Approach. Lippincott Williams & Wilkins, Philadelphia, 2005.		
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Learning outcomes 1 and 6 are tested during experimental work in laboratory practicals and tests, outcome 7 during seminars, and outcomes 2-5 with written and oral exam.		